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Metabolism and carbonate buffering drive seasonal dynamics of CO₂ emissions from two German reservoirs

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Abstract

Biological metabolism drives much of the variation in CO₂ in terrestrial ecosystems but does not explain CO₂ oversaturation and emission in net autotrophic lakes and reservoirs. The unexplained CO₂ could be attributed to the equilibria between CO₂ and the carbonate buffering system, which is seldom integrated into CO₂ budgets, let alone its interplay with metabolism on CO₂ emissions. Here, we perform a process-based mass balance modeling analysis based on an 8-year dataset from two adjacent reservoirs with similar catchment sizes but contrasting trophic states and alkalinity. We find that in addition to the well-acknowledged driver of net metabolic CO₂ production, carbonate buffering also determines the total amount and seasonal dynamics of CO₂ emissions from the reservoirs. Carbonate buffering can contribute up to nearly 50% of whole-reservoir CO₂ emissions, by converting the ionic forms of carbonate to CO₂. This results in similar seasonal CO₂ emissions from reservoirs with differing trophic state, even in low alkalinity system. We therefore suggest that catchment alkalinity, instead of trophic state, may be more relevant in predicting CO₂ emissions from reservoirs. Our model approach highlights the important role of carbonate buffering and metabolism that generate and remove CO₂ throughout the reservoirs on a seasonal scale. The inclusion of carbonate buffering could diminish a major uncertainty in the estimation of reservoir CO₂ emissions and increase the robustness of aquatic CO₂ emission estimates.

Introduction

Carbon dioxide (CO₂) is the dominant gaseous end product of the C cycling and the largest contributor to the anthropogenic greenhouse effect (IPCC, 2022). Damming of lotic water, which is essential for human welfares, deeply affects the freshwater carbon footprint by facilitating internal processing with longer water residence time. Reservoirs are typically supersaturated in CO₂ relative to atmosphere (Deemer et al., 2016; Stets et al., 2009), making them a CO₂ source, while some studies suggest that eutrophic reservoirs and lakes can turn into CO₂ sinks due to increases in aquatic primary productivity (Bogard and Del Giorgio, 2016; Jones et al., 2016). Reservoirs are responsible for substantial increases of the global lake area (Pi et al., 2022), highlighting the critical need for precise quantification of CO₂ emissions from reservoirs. Currently, net ecosystem production (NEP) is usually considered as the main driver of CO₂ supersaturation in reservoirs, thus trophic states, as the indicator of NEP, is widely applied as a robust proxy to estimate global lake and reservoir CO₂ emissions (DelSontro et al., 2018; Kortelainen et al., 2006; Trolle et al., 2012). However, recent observations of CO₂ dynamics in rivers and lakes indicate that net autotrophic systems also emit CO₂ (Bogard and Del Giorgio, 2016; McDonald et al., 2013; Stets et al., 2017), suggesting that metabolism alone is not sufficient in explaining CO₂ emissions in lakes and reservoirs (Marcé et al., 2015; Weyhenmeyer et al., 2015). The underlying processes that sustain CO₂ emissions from the productive reservoirs are, however, still far from being understood.

Whether a given system emits or consumes CO₂ depends on the net balance of external inputs/outputs and in-reservoir production/consumption via biological and chemical processes. All carbon entering reservoirs can be converted to CO₂ via three distinct pathways (**Fig. 1a**): 1) direct CO₂ input from external inflows (inflows, lateral soils, and groundwater), 2) biotically via net metabolic CO₂ production within reservoir, and 3) abiotically via carbonate buffering (Duvert et al., 2019; Stets et al., 2017). Direct external CO₂ can be derived from terrestrial organic carbon (OC) mineralization and catchment weathering, fueling intense aquatic CO₂ emission to the atmosphere (Duvert et al., 2019; Weyhenmeyer et al., 2015; Xiao et al., 2020). Net metabolic CO₂ production is the sum of all biotic processes that convert other forms of carbon (particulate organic carbon, POC; dissolved organic carbon, DOC; particulate inorganic carbon, PIC) to dissolved inorganic carbon (DIC), including whole-reservoir respiration, photosynthesis, anaerobic respiration, and photochemical DOC mineralization. In reservoirs with high DOC concentrations where external DOC input is the major contributor, higher net heterotrophy (respiration exceeding photosynthesis) is usually observed (Alleson et al., 2021; Bogard and Del Giorgio, 2016; Lapierre et al., 2013). In contrast, increased terrestrial nutrient delivery is known to stimulate in-

lake productivity by enhancing growth of phototrophs, leading to higher CO₂ consumption (Li et al., 2021).

Although seasonal patterns in net metabolic CO₂ production have been widely observed (Trolle et al., 2012; Vachon et al., 2017), the external derived CO₂ from carbonate buffering in supporting reservoir emissions, however, is seldom distinguished, especially regarding their variations in time (Vachon and Del Giorgio, 2014). External inorganic carbon inputs vary seasonally due to dynamic hydrologic regimes and soil respiration in the catchments (Chaplot and Mutema, 2021; Tittel et al., 2015). Meanwhile, the carbonate equilibrium also strongly modulates CO₂ in reservoirs based on the speciation of DIC (Marcé et al., 2015). In freshwaters with high pH/alkalinity, CO₂ is in equilibrium with the ionized forms, HCO₃⁻ and CO₃²⁻ (Stets et al., 2017), decoupling CO₂ dynamics from internal metabolism. Carbonate buffering (by converting dissolved CO₂ into ionic forms, and vice versa), therefore, may contribute a large fraction of CO₂ evading to the atmosphere via large inorganic carbon import from catchments.

Given the seasonality of nutrient loads from catchments (Leng et al., 2021; Tittel et al., 2015), we assume lower metabolic CO₂ production (net autotrophy) and higher external DIC inputs partially offset each other, resulting in more stable CO₂ emissions from eutrophic reservoirs (**Fig. 1b**). Primary production, enhanced by increases in phytoplankton productivity, depletes CO₂ and enhances pH in nutrient-rich reservoirs during summer (Li et al., 2021). When external DIC loading from catchments contributes to the system, DIC input or removal will change carbonate equilibrium and CO₂ fluxes. However, little is known about the extent to which carbonate buffering contributes to CO₂ evasion on a seasonal scale. Particularly, when alkalinity is low (0-1 meq L⁻¹), metabolic CO₂ production has been thought to be the major driver of CO₂ dynamics (Khan et al., 2020). It is still unclear whether carbonate buffering has the potential to reverse CO₂ undersaturation to supersaturation, bringing large uncertainty in CO₂ emission estimates.

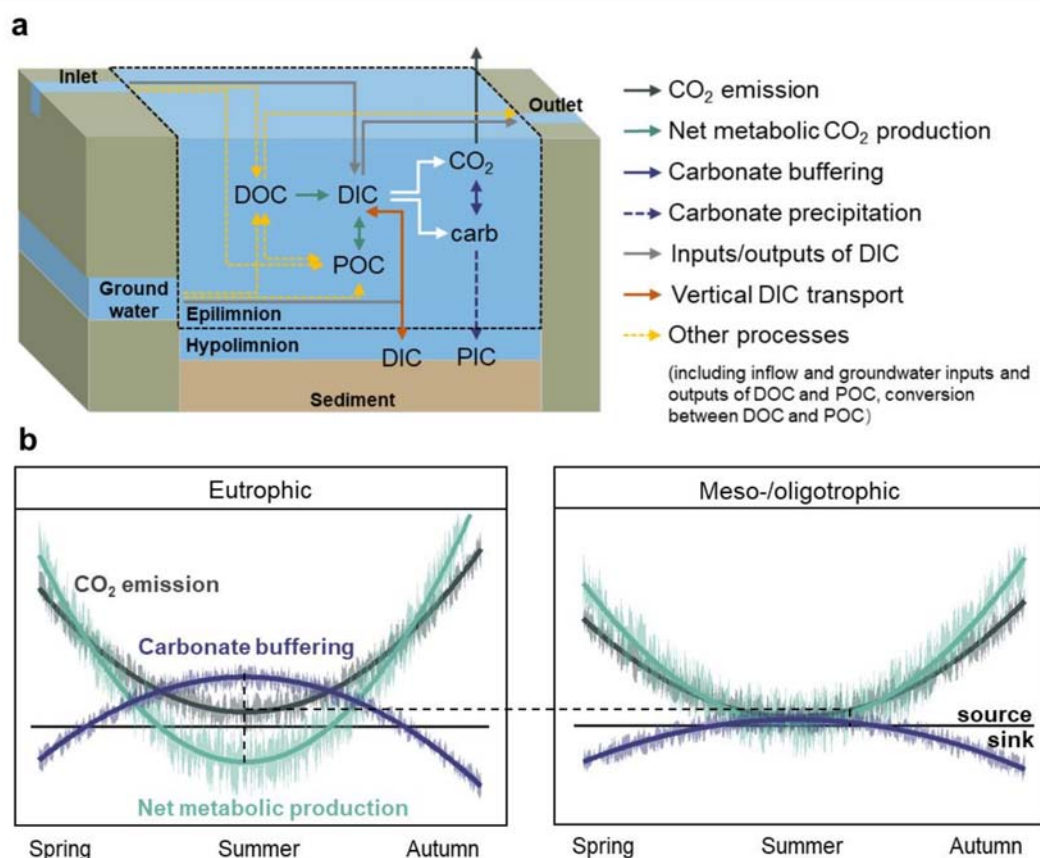


Figure 1. Model diagram and hypotheses. **a.** Scheme of C sources, forms, and biogeochemical transformations in epilimnion for the stratified reservoir systems or entire water column for the unstratified systems. The dashed black line represents the model boundary. The white arrows represent that dissolved organic carbon (DIC) consisting of carbonates (indicated by "carb", $\text{HCO}_3^- + \text{CO}_3^{2-}$) and CO_2 . DOC, dissolved organic carbon; POC, particulate organic carbon; PIC, particulate inorganic carbon. **b.** Hypothetical responses of CO_2 emissions to seasonal-dependent processes in reservoirs with contrasting productivity and alkalinity. The seasonal change in CO_2 emissions is governed by the balance of in-reservoir processes and external inputs: net metabolic CO_2 production (green lines) and carbonate buffering (purple lines). Both processes can produce and consume CO_2 biotically and abiotically, respectively. The dashed black lines show different amounts of net metabolic CO_2 production and carbonate buffering from reservoirs with distinct trophic states result in the same CO_2 emissions.

Net metabolic CO_2 production and carbonate buffering are typically studied by comparison between rate of changes in O_2 and DIC (Khan et al., 2020; Marcé et al., 2015; Stets et al., 2017). This method interprets that the deviations of the theoretical slope (-1) between DO and DIC disequilibrium are caused by carbonate precipitation, therefore it largely overlooks: (1) the varying

respiratory/photosynthetic quotient (Trentman et al., 2023); (2) the effects of anoxic and photochemical processes, that potentially alter the correlation between DIC and O₂; and (3) low-alkalinity systems with seasonal fluctuation in the case of DIC inputs/removals. Biological net metabolic CO₂ production rates in reservoir are often assessed using diel measurements of free-water dissolved O₂ capturing metabolic dynamics (Winslow et al., 2016). To evaluate the role of all processes which may govern CO₂ dynamics in the reservoir ecosystems, we established a coupled DIC and CO₂ mass balance using in-situ measures of DIC and CO₂ at a temporal resolution that captures seasonal variation. The estimates of net metabolic CO₂ production can be achieved by the balance between CO₂ emissions and the difference between DIC inputs and outputs. By contrasting the DIC mass balance with the CO₂ one, we are able to quantify the effect of carbonate buffering on reservoir estimates of CO₂ processing.

The aim of this study is to disentangle the seasonal patterns of CO₂ dynamics and to quantify the relative contribution of the key underlying biotic and abiotic processes that drive seasonal pattern in CO₂ emissions. To reconstruct the seasonal dynamics of these processes supporting CO₂ emission, we selected two adjacent temperate reservoirs with contrasting trophic states (eutrophic and mesotrophic) and collected samples at biweekly interval for more than 8 years. We hypothesized (1) a seasonal succession in the relative contributions of each process influencing reservoir CO₂ budgets (only epilimnion during stratification), and (2) eutrophication accompanied by external inorganic carbon loading resulting in attenuated seasonal variation in CO₂ emissions. We built a simple mass balance model coupling reservoir DIC and CO₂ budgets using the routine monitoring data to realistically interpolate rates of different pathways supporting CO₂ emissions. In addition, we validated our mass balance approach by comparing the net metabolic CO₂ production with the NEP estimated from free-water oxygen measurements. Finally, we developed a statistical model to explore how sensitive reservoir CO₂ dynamics may be to climatic drivers (flow discharge and sunlight duration of the day) and major biotic and abiotic processes.

Methods

Stream-reservoir systems

The Hassel and Rappbode catchments are located in the eastern Harz Mountains in Central Germany. Both catchments share similar catchment size and morphological properties (40.5 and 39.1 km² for Hassel and Rappbode, respectively). The Hassel catchment is more affected by agriculture (37% forest and 25% agricultural fields) while Rappbode is primarily forested (72% forest and 2% agricultural fields, **Fig. S1**) (Friese et al., 2014). This explains the distinct trophic states in Hassel (eutrophic, **Fig. S1**) and Rappbode (mesotrophic)

reservoirs (Kong et al., 2022). The inflow streams Hassel and Rappbode drain directly into the pre-dam reservoirs Hassel and Rappbode, respectively (referred to as "Hassel Reservoir" and "Rappbode Reservoir" hereafter). The reservoirs were built to reduce sediment and nutrient load to the main reservoir of Rappbode, which is the largest drinking water reservoir in Germany. The Hassel and Rappbode reservoirs were brought into operation in 1960 and 1961, respectively (Wouters, 2011). Both reservoirs are configured with surface-release dams, resulting in constant water depth (**Fig. S1**). Thus, we assumed the same concentrations of solutes in surface reservoir water and outflow. Monitoring stations were located at the inflow streams and the reservoir surface close to the dam. As the water level in reservoirs in the dataset varied within 0.3 m in the studied years, we reasonably assumed that the surface area of reservoirs remained constant (0.26 km² in Hassel; 0.243 km² in Rappbode, Halbedel and Koschorreck (2013)).

Data collection

Biweekly measurements of water temperature, pH, alkalinity, total inorganic carbon (TIC), DOC, and total phosphorus (TP) were carried out from stream and reservoir monitoring sites for physical and chemical analysis in the field and laboratory from 2014 to 2021 (including ice-covered period). The lab methods and data handlings were described by Friese et al. (2014) in more detail.

Discharge data of both inflow streams were logged with 15-min intervals at the official automatic gauging station. At Hassel stream, the gauge is located 2 km upstream of the sampling site (51°41'54.0"N, 10°50'24.6"E). We corrected discharge assuming equal precipitation within the whole catchment (Wendt-Potthoff et al., 2014). The daily discharge of outflows of the pre-dam reservoirs was provided by the local dam operation company (Talsperrenbetrieb Sachsen-Anhalt, TSB). Meteorological data, including air pressure (hPa), air temperature (°C), precipitation (mm), wind speed (m s⁻¹), and solar radiation (J cm⁻²) were collected at hourly interval from a weather station at Rappbode dam for both pre-dam reservoirs (weather station at YRA, see Friese et al. (2014)). Meteorological and discharge data were aggregated to daily mean values to match with other monitoring data for further analysis. The water residence time is defined as the ratio between the water volume of the reservoir and the daily inflow.

For CO₂ concentration, 30 mL of water samples were equilibrated with 30 mL N₂ or surrounding atmosphere in a gas tight syringe through vigorous shaking. N₂ was used as headspace gas before 2020 and the equilibrated gas was injected directly into the gas chromatograph (GC, SRI 8610, SRI Instruments, Torrance, USA). Since then, ambient air was used as headspace gas and the equilibrated gas was stored in evacuated Exetainer vials (Labco, UK) for later GC analysis

equipped with an auto sampler. Our previous work showed that both N₂ and ambient atmosphere as headspace gas generated identical CO₂ concentrations (Koschorreck et al., 2021). Ambient CO₂ concentrations and alkalinity were separately measured to correct the headspace CO₂ results due to the chemical equilibration of the carbonate system in syringes. The method can be implemented by using an R script provided by Koschorreck et al. (2021). Detailed methods for CO₂ analysis and calculation can be found in Leng et al. (2021). DIC was calculated from measured alkalinity, temperature, and *p*CO₂ by using the R package *seacarb* (Gattuso et al., 2021) using salinity constants of zero given the electrical conductivity < 400 μS cm⁻¹ (at 25 °C) in all observations.

To model the dynamics of net ecosystem production (NEP), high temporal resolution measurements of physiochemical parameters were collected. We deployed an in-situ Zebra D-OPTO sonde (YSI6800, YSI, USA) suspended at 0.5m depth below the water's surface to measure water temperature and dissolved oxygen (O₂) at 15-min intervals. The O₂ logger was deployed from October 2015 to May 2016, July 2017 to December 2018, and March to December 2021. The O₂ logger was calibrated using 100% saturated O₂ water before and after deployment. Data were drift-corrected assuming a constant drift of the calibration during deployment. O₂ and CO₂ departure from atmospheric equilibrium (ΔO₂ and ΔCO₂) was calculated as the difference of O₂ and CO₂ concentrations in water and atmosphere.

Determination of CO₂ emissions

Flux rates (mmol CO₂ m⁻² d⁻¹) were determined as the product of gas transfer velocities (*k*_{CO₂}, m d⁻¹) and CO₂ concentration gradients (μmol L⁻¹) between water and air. The equations for *k*_{CO₂} calculation in reservoirs are provided in *Supporting information Table S1*. *k* was estimated by the relationship between *k*₆₀₀ and daily wind speed according to Crusius and Wanninkhof (2003). The *k*₆₀₀ was then converted to a gas- and temperature-specific gas transfer velocity *k*_{CO₂} using the temperature dependent Schmidt numbers for CO₂ (Crusius and Wanninkhof, 2003). We estimated the annual CO₂ flux (Mg C y⁻¹) for the pre-dam reservoirs Hassel and Rappbode by multiplying the mean surface area of the reservoirs by monthly averaged CO₂ fluxes.

Coupled CO₂-DIC modeling

We developed a simple process-based mass balance model based on the conservation of mass of CO₂ to estimate net metabolic CO₂ production and carbonate buffering capacity within the stream-reservoir system (**Fig. 1a**). The model boundary represents a completely mixed epilimnion (for stratified reservoirs) or unstratified reservoir. Changes of CO₂ between outlet and inlet are

theoretically caused by groundwater exchange, net metabolic CO₂ production (NMP, positive: production; negative: consumption), carbonate buffering (CB, positive: conversion of carbonates into CO₂; negative: conversion of CO₂ into carbonates), vertical CO₂ input (positive: vertical diffusion from deeper layers, including the input from hypolimnion to epilimnion during stratification and from sediment to mixed water column during mixing), and CO₂ emissions. The DIC mass balance can be calculated by accounting for changes in DIC pools resulting from groundwater exchange, NMP, vertical DIC input, and CO₂ emissions within reservoir. The biweekly mass balance model was run and the daily rates of different processes were smoothed by fitting multiple regressions in local neighborhood (loess). Process rates were presented as absolute daily rates (mmol m⁻² d⁻¹) for parallel comparison by dividing the surface area of reservoirs. The mass balance model allows us to differentiate the carbonate buffering from potential in-reservoir metabolic production of CO₂, and therefore their relative contributions to CO₂ emissions. We parameterized the model using our own measurements and compiled all equations to solve for processes in CO₂ and DIC mass balance (**Table 1**). The detailed model parametrization processes are provided in *Supporting information Text S1*.

Table 1. Processes rates (mol d⁻¹) expressed in equations in the CO₂ and DIC mass balance model, where Q_{in} and Q_{out} are the discharge at the inlet and outlet, respectively. [CO_{2in}] and [CO_{2out}] are the CO₂ concentrations at the inlet and outlet, respectively. [DIC_{in}] and [DIC_{out}] are the DIC concentrations at the inlet and outlet, respectively. MLD represents the mixing layer depth. ΔVol is the volume of water transferred between hypolimnion and epilimnion. [DIC_{hypo}] and [DIC_{epi}] are the DIC concentrations at the hypolimnion and epilimnion, respectively. F_{CO2} is the CO₂ flux between air-water interface. A is the surface area of the reservoir. The detailed model parametrization are described in *Supporting information Text S1*.

Parameters	Symbol	Equation
Inflow CO ₂ and DIC loads	CO _{2, in} and DIC _{in}	[CO _{2in}] × Q _{in} and [DIC _{in}] × Q _{in}
Outflow CO ₂ and DIC loads	CO _{2, out} and DIC _{out}	[CO _{2out}] × Q _{out} and [DIC _{out}] × Q _{out}
Vertical input of DIC	DIC _v	MLD increase: ΔVol × [DIC _{hypo}] MLD increase: ΔVol × [DIC _{epi}]
CO ₂ emission	E	F _{CO2} × A
Net metabolic CO ₂ production	NMP	[DIC _{out}] × Q _{out} - C _{DIC, in} × Q _{in} + F _{CO2} × A
Carbonate buffering	CB	(([DIC _{in}] - [CO _{2in}]) × Q _{in} - ([DIC _{out}] - [CO _{2out}]) × Q _{out})

We made several simplifying assumptions in constructing the model. Vertical input can add or remove considerable DIC (and CO₂) to the DIC budgets,

especially during autumn turnover (Vachon et al., 2017). The diffusive exchange of solutes between epilimnion and hypolimnion in both studied reservoirs, however, is typically orders of magnitude slower than processes in the epilimnion (Kreling et al., 2014) and the vertical entrainment of DIC and CO₂ (DIC_v and CO_{2,v}) should be negligible compared to other biological and hydrological processes. We quantified more thoroughly this assumption by evaluating DIC mass transport (water column mixing) between layers when mixing layer changed (see *Results* and *Supporting information* Text S2 for further details). It should be noted that neglecting DIC_v would only affect estimates of net metabolic CO₂ production in our model, not carbonate buffering. We further calculate the potential role of vertical input on the mass balance model by comparing estimated NMP changes with and without the effect of vertical input.

We assumed that the contributions of CO₂ and DIC from groundwater can be neglected in our systems because the local geological units are covered with thin alluvium layers that are too thin to develop proper aquifers (Barth et al., 2017). The role of methane was excluded because its concentration and fluxes were generally three orders of magnitude lower than CO₂ (Tittel et al., 2019). To verify whether DIC converted to PIC (calcite precipitation) during carbonate buffering within the system, we estimated calcite saturation indices (SI) with measured concentrations of Ca²⁺ and HCO₃⁻ by using PHREEQC (Parkhurst and Appelo, 2013). For samples with higher SI (SI > 0), thermodynamics will favor calcite precipitation with 2 mol of carbonates producing 1 mol each of CaCO₃ and CO₂. SI within the system ranged from -2.32 to -0.77 with the mean value of -1.56, implying calcite precipitation was not likely to occur in our reservoirs.

Ecosystem metabolism

We compared the NMP from mass-balance modeling with net ecosystem production (NEP) calculated from diel O₂ data using the LakeMetabolizer R package (Winslow et al., 2016). Reservoir NEP is the balance of gross primary production (GPP) and ecosystem respiration (ER), both of which influence diel oxygen concentration. We used the Bayesian model from Holtgrieve et al. (2010) considering hourly time series of dissolved oxygen, water temperature, wind speed, mixing layer depth (MLD), and photosynthesis active radiation (PAR). The MLD was calculated based on modeled water temperature profiles at daily time step throughout the study period in these two studied reservoirs (Kong et al., 2022), using the R package *RLakeAnalyzer* (Winslow et al., 2019). Missing values of mixing layer depth were checked for unstratified occurrences, then we used the depth of the reservoirs as the mixing layer depth instead. PAR (μmol m⁻² s⁻¹) was calculated from solar radiation (J cm⁻²) using a conversion factor of 4.56 μmol J⁻¹. We fitted the model resulting in estimates of GPP, ER, and NEP for each

day. Here, we multiplied the generated volumetric unit by mixing layer depth to report the estimates in area units ($\text{mmol O}_2 \text{ m}^{-2} \text{ d}^{-1}$) for ease of comparison with the results from mass-balance modeling. NEP values were converted to units of CO_2 assuming a quotient of 1 (mol C/mol O). We then matched the NEP data with NMP from mass-balance model based on the measured data.

Statistical analysis

We applied piecewise structural equation modeling (SEM) to interpret the causal and directional links between physical and biological parameters operating on CO_2 emissions. We constructed a network to simultaneously evaluate the climatic drivers and different pathways affecting reservoir CO_2 . The network consists of two dependent layers of factor interaction: the first one is whether CO_2 emissions arise from environmental variables; the second one is whether the parallel variations of net metabolic CO_2 production and carbonate buffering determined the CO_2 emissions. To estimate the predictive strengths in both reservoirs, we fitted models for both reservoirs using the function *psem* in the R package *piecewiseSEM* (Lefcheck, 2016). The estimates of standardized path coefficients were reported for evaluating the effect sizes of each relationship (or path). The evaluation of goodness of fit and associated uncertainty was performed through the coefficient of determination (R^2) and the residual standard error (RSE), respectively. The overall fit of the model was evaluated using Shipley's test of d-separation: Fisher's C statistic (if $p > 0.05$, then no paths are missing and the model is a good fit). All modeling calculation and statistical analyses were performed in R version 3.5.1 (R Core Team, 2021).

Results

Physical and chemical characteristics of reservoirs

Overall, both reservoirs had similar seasonality, with Hassel having higher water residence time (WRT), alkalinity, DIC, DOC, and TP (**Fig. 2**, Wilcoxon test, $p \leq 0.001$). WRT in both reservoirs was driven by changes in discharge of inflows since the volumes of both reservoirs remained constant. During summer when both reservoirs were thermally stratified, decreases in inflow discharge were followed by increases in WRT (**Fig. 2a** and **Fig. S2a**). ΔO_2 ranged from -200 to $180 \mu\text{mol L}^{-1}$ with a median of $-40 \mu\text{mol L}^{-1}$ in both reservoirs. Reservoirs were mostly undersaturated in O_2 with the exception of summer (**Fig. 2b**). Both reservoirs were characterized by longer residence time and higher nutrients in summer (**Fig. 2a, c, d**). The seasonality of alkalinity and TP (lower in early spring and higher in autumn) followed the opposite pattern as nitrate. pH values shared similar seasonal variations with alkalinity, ranging between 6.5 and 9.8 with 7.5

as the median. NO_3 , alkalinity, and DIC in both reservoirs exhibit similar seasonal patterns to inflows (**Fig. S2**).

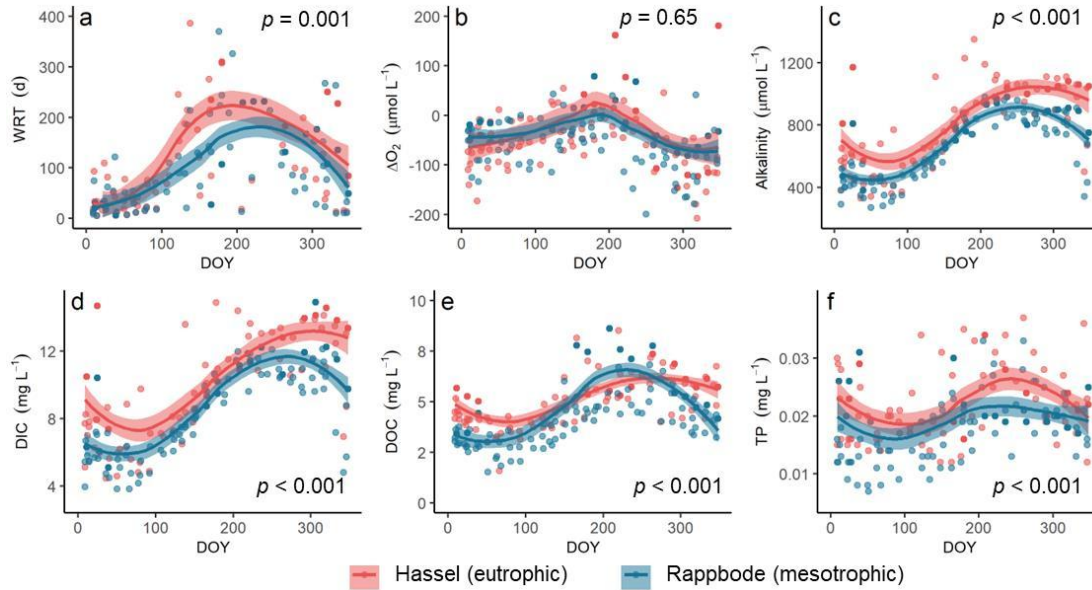


Figure 2. Seasonal variations in water residence time (WRT, a), O_2 departure from atmospheric equilibrium (ΔO_2 , b), alkalinity (c), dissolved inorganic carbon (DIC, d), dissolved organic carbon (DOC, e), and total phosphorous (TP, f) in Hassel and Rappbode Reservoirs, respectively. Dots are the point measurements during the 8-year monitoring. Solid lines with shaded area in the time series plots are locally weight regression model fittings (loess) for visual interpretation of the temporal dynamics. The shaded area represents 95% confidence intervals. p values represent the significance of differences between both reservoirs by Wilcox tests. DOY: day of the year.

Both reservoirs were mainly supersaturated with CO_2 (**Fig. S3**). CO_2 was oversaturated in 232 out of 264 observations, with a median of $44 \mu\text{mol L}^{-1}$. Additionally, ΔO_2 and ΔCO_2 for both reservoirs significantly negatively correlated with each other ($p \leq 0.001$). Both reservoirs showed an offset relative to the expected -1:1 line with larger offset in Hassel (200 and $90 \mu\text{mol L}^{-1}$ in Hassel and Rappbode according to the method from Vachon et al. (2020)). The slope of the linear regressions between ΔO_2 and ΔCO_2 for both reservoirs were lower than -1 (estimate = -1.21 , $R^2 = 0.27$, $p < 0.001$).

Patterns of CO_2 emissions

On average, both reservoirs were net sources of CO₂ to the atmosphere, despite a few periods of CO₂ uptake in summer (**Fig. 3**). Inflows showed a clear difference in magnitude and seasonal variation of CO₂ concentrations compared to reservoirs (**Fig. 3**). Median CO₂ concentrations of the Hassel and Rappbode reservoirs were 22 and 23 µmol L⁻¹, respectively, which were much lower than 74 and 50 µmol L⁻¹ in both inflows. CO₂ emission rates of reservoirs ranged from -28 to 348 mmol m⁻² d⁻¹, with median values of 28 and 35 mmol m⁻² d⁻¹ in Hassel and Rappbode, respectively. Apart from the apparent seasonal variations, we did not find significant monotonic long-term trends in CO₂ emissions (Mann-Kendall test, $p > 0.05$). Given the ice-free period from April to December, annual mean CO₂ emissions from Hassel and Rappbode Reservoirs were estimated at 16.2 and 16.5 Mg C yr⁻¹, respectively. Both concentrations and emissions in reservoirs were higher in winter and lower in summer. Contrasting to significant difference of CO₂ concentrations between both inflows, CO₂ concentrations (t-test, $p = 0.27$) and emissions (t-test, $p = 0.56$) in the two reservoirs with divergent trophic states were not significantly different from each other.

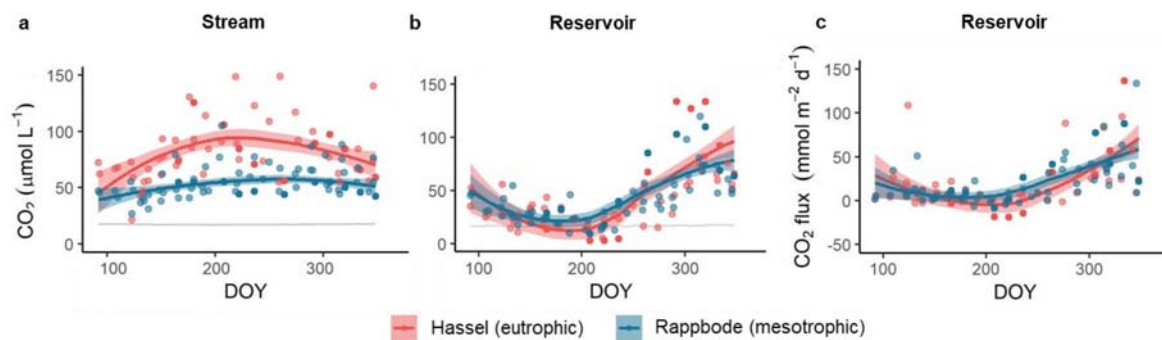


Figure 3. Seasonal variations in the CO₂ concentrations in the Hassel and Rappbode inlets (a) and Reservoirs (b) and fluxes in both reservoirs (c) and from April to December (full ice-covered period is excluded). Dots are the point measurements during the 8-year monitoring. The grey lines represent the atmospheric equilibrium concentration of CO₂. DOY: day of the year.

Given k_{CO_2} and CO₂ concentration directly drive CO₂ emissions, we observed higher dependency between emissions and concentrations than between emissions and k_{CO_2} in Hassel (the inclusion of k_{CO_2} enhances the predictive power of linear regression model from 70% to 92%), whereas similar contribution of CO₂ concentrations and k_{CO_2} to the variance of CO₂ emissions were observed in Rappbode (the predictive power of linear regression model increases from 41% to 91%). Regarding the relationships between CO₂ emissions and other environmental indices (**Fig. S4**), light (sunlight duration of the day, $p < 0.001$, $R^2 = 0.21$, $n = 262$) and pH ($p < 0.001$, $R^2 = 0.25$, $n = 262$) have stronger negative relationships with CO₂ emissions than did water temperature ($p < 0.001$, $R^2 = 0.15$, $n = 262$) and WRT ($p < 0.001$, $R^2 = 0.14$, $n = 256$). Compared to above

correlations, the positive link between nitrate and CO₂ emissions was rather moderate ($p < 0.001$, $R^2 = 0.05$, $n = 262$).

Mass balance modeling of CO₂ budgets

Mass balances of DIC and CO₂ were explored to provide evidence for the contributions of different processes supporting CO₂ evasion in reservoirs. Contributions of inflows to the CO₂ fluxes (mmol m⁻² d⁻¹) in mass balance budget were close to 0 for the whole period (1st-3rd quartile: 1.3-6.0 mmol m⁻² d⁻¹, **Fig. 4a, b**). That means that equal amounts of CO₂ entering the reservoir at the inflow also left the reservoir at the outflow.

Compared to the contribution of inflow (**Fig. S5**, median: 5.9% in Hassel and 4.2% in Rappbode), NMP and carbonate buffering were the primary contributors to net CO₂ emissions in both reservoirs (**Fig. S5**, median: 46% and 45% in Hassel; 57% and 39% in Rappbode). NMP varied remarkably following similar temporal patterns in both reservoirs, which decreased below 0 in summer (GPP>ER) and became positive again from autumn onwards. However, the variation range of NMP in Hassel was lower than in Rappbode (**Fig. 4c, d**). The variation in NMP (1st-3rd quartile: -19.0-42.0 mmol m⁻² d⁻¹) in both reservoirs was larger than the one in CO₂ emissions (1st-3rd quartile: 5.6-30.0 mmol m⁻² d⁻¹).

Carbonate buffering had opposite fluctuations compared to NMP with narrower variation ranges (1st-3rd quartile: -1.5-27.8 mmol m⁻² d⁻¹, **Fig. 4c, d**), especially in the Rappbode Reservoir. As a result, despite NMP showed the possibility of CO₂ deficit in the systems, CO₂ mostly sustained outgassing during summer (CO₂ emissions > 0, **Fig. 4e, f**) due to the contribution of carbonate buffering. Due to lower alkalinity in Rappbode, CO₂ emissions were moderately modulated by carbonate buffering. Consequently, different variations of NMP and carbonate buffering between both reservoirs shaped similar temporal patterns of CO₂ emissions.

A negative correlation between the influence of NMP and carbonate buffering was observed with similar slopes (**Fig. 5a**, CB ~ NMP, Hassel: slope = -0.80, $p < 0.001$, $R^2 = 0.82$; Rappbode: slope = -0.89, $p < 0.001$, $R^2 = 0.89$). Regarding the median contributions of different sources over years, NMP and carbonate buffering were equally important to CO₂ dynamics in Hassel Reservoir, whereas there was a stronger influence of NMP on CO₂ dynamics in Rappbode Reservoir. All in all, CO₂ emissions were majorly driven by in-reservoir metabolic production and DIC import from the catchment.

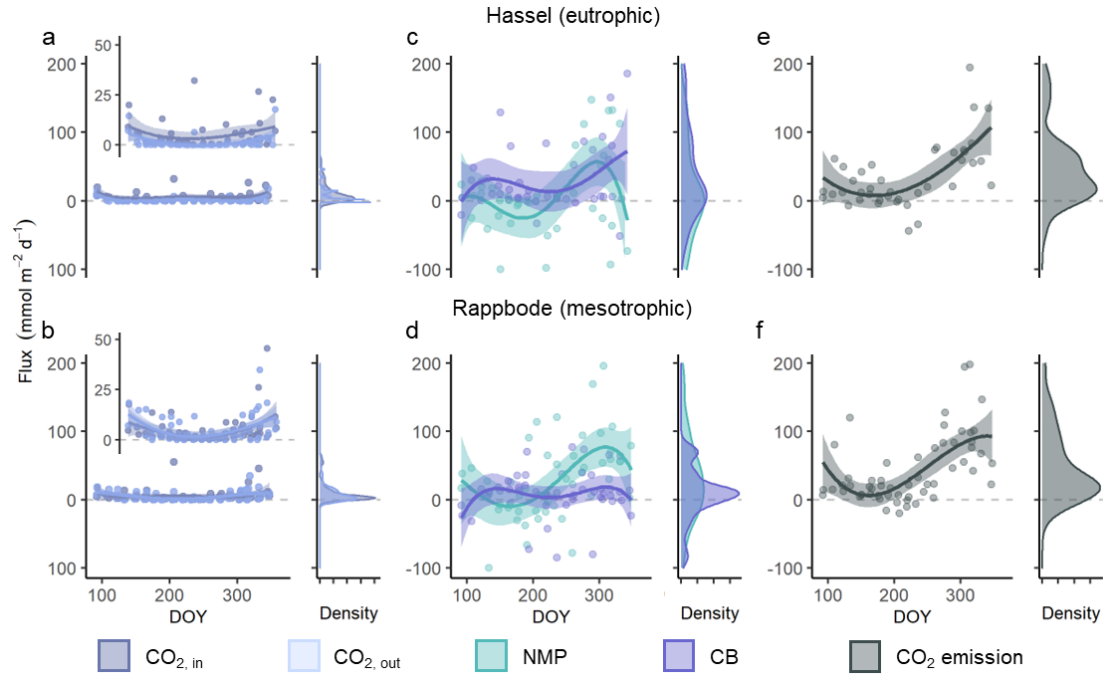


Figure 4. Season variations and density distributions of inlet input and outlet export flux of CO₂ (a and b), net metabolic CO₂ production (NMP) and carbonate buffering (CB, c and d), and CO₂ emissions (e and f) in areal CO₂ flux rates (mmol m⁻² d⁻¹) from the results of coupling DIC-CO₂ mass-balance modeling in both reservoirs. Solid lines with shaded area in the time series plots are locally weight regression model fittings (loess) for visual interpretation of the temporal dynamics. The shaded area represents 95% confidence intervals. Fluxes of inputs and exports were normalized by dividing them by the surface area of the reservoir for comparison with other processes. The inner panels in a and b show the same seasonal variations in inflow input and outflow export in both reservoirs with a smaller scale. DOY: day of the year.

If the mixed layer depth increases, DIC rich hypolimnetic water is mixed into the epilimnion, increasing the epilimnetic DIC pool. This has an impact on NMP in our mass balance. We evaluated the potential influence of vertical DIC transfer on NMP in our model by multiplying DIC concentration by the volume of the water layer exchanged between epilimnion and hypolimnion (**Table 1**, see Supporting information **Text S2** for more details). Compared with NMP, the gain or release of hypolimnetic DIC was rather low at both reservoirs over the seasonal cycle except during autumn mixing (consistently below 5% and 12% of CO₂ produced or consumed via NMP in Hassel and Rappbode reservoir during 80% of the time, respectively, **Fig. S6**). During autumn turnover, some DIC may have been transported from the hypolimnion to the epilimnion. During this period, vertical input can account for up to 24% of estimated NMP to the upper layer in both reservoirs (**Fig. S6b, c**). Direct import of CO₂ from the hypolimnion was also limited due to the neutral pH in both reservoirs (Dadi et al., 2023), which means

that the majority of DIC was bicarbonate rather than CO_2 . As a result, we conclude that neglecting vertical input during mixing surely induced uncertainties to the modelled NMP rates. This means that NMP cannot be equaled with -NEP because a small part of the epilimnetic DIC pool change was resulting from reservoir turnover. This is, however, not affecting our quantification of carbonate buffering.

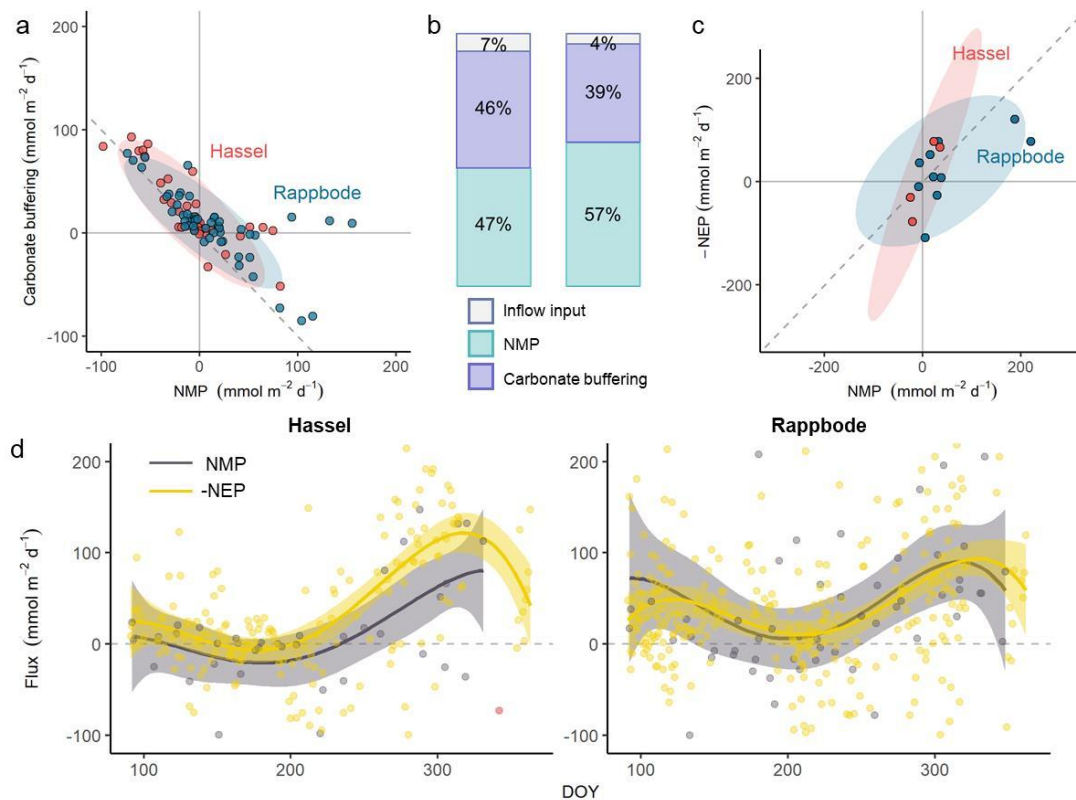


Figure 5. a. Correlations between carbonate buffering and net metabolic CO₂ production (NMP) in Hassel (eutrophic) and Rappbode (mesotrophic) reservoirs during the ice-free period. b. Median contributions of different processes to CO₂ emissions from the results of coupling DIC-CO₂ mass-balance modeling in both reservoirs. c. Comparison between NMP from mass-balance modeling and net ecosystem productivity (NEP) modeled from high-frequency O₂ data. The dashed grey line is the 1:1 line, where the internal productivity from both methods perfectly matches. d. Net metabolic CO₂ production expressed in areal CO₂ rates estimated by mass balance calculations (grey dots) and whole-reservoir net ecosystem metabolism (-NEP) derived from free-water oxygen measurements (yellow dots). Seasonal smoothed trends are calculated using a fourth-degree polynomial regression for net metabolic production (grey line) and -NEP (yellow line). The red dot as the outlier in the Hassel Reservoir was excluded for generating the smoothed trend of NMP to eliminate the effect on the smoothed line. Note here that positive values represent CO₂ production (net heterotrophy). DOY, day of the year.

To evaluate the reliability of the results from mass balance modeling, we estimated the -NEP by modeling the diel O₂ variation in the reservoir systems. The smoothed estimates of NMP from the mass balance model and -NEP from the lake metabolism model followed the same pattern across seasons (**Fig. 5d**). NMP and -NEP corresponded to the epilimnetic CO₂ concentrations during the stratified period. The offset between the two remained relatively conservative during summer.

Controls of CO₂ emissions

To infer the direct and indirect effects of light (sunlight duration), inflow discharge, DIC, and DOC with causal combination that underpin variations in CO₂ emissions we fitted a piecewise SEM model. The model explained 65% of the variation in CO₂ emissions from both reservoirs (**Fig. 6**). We found direct support for a negative relationship between light and NMP, and we did observe that lower inflow discharge was correlated with higher inflow DIC and in-reservoir DOC. CO₂ emissions were highest when NMP, carbonate buffering, and inflow CO₂ load were high. Variation in CO₂ emissions was primarily related to the patterns of NMP and carbonate buffering. While NMP also had negative effects on carbonate buffering, indicating both processes constrained each other.

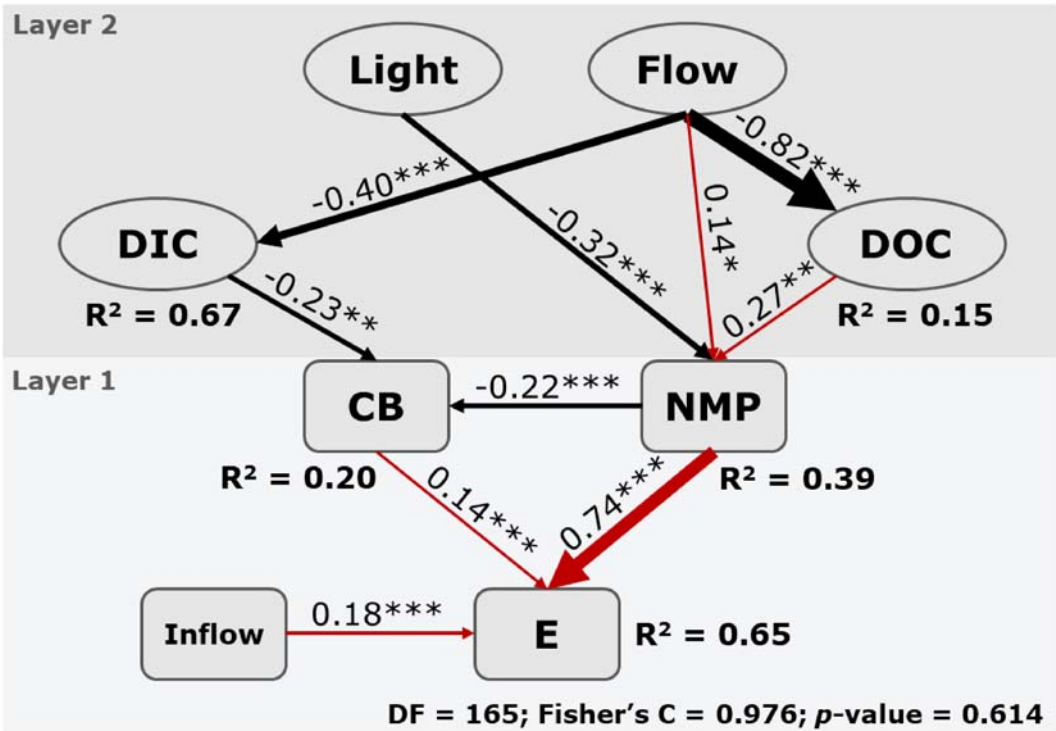


Figure 6. Structural equation model (SEM) representing connections between reservoir physical and biological parameters contributing to seasonal variations in CO₂ emissions (E). The SEM consisted of two dependent levels of factor interactions. Layer 1 assessed the influence of net metabolic CO₂ production (NMP), carbonate buffering (CB), and inflow CO₂

load on CO₂ emissions from reservoirs. Layer 2 assessed relationships between environmental variables (Light: sunlight duration of the day; Flow: inflow discharge; DIC: inflow DIC concentration; DOC: reservoir DOC concentration) and variations in CO₂ emissions. Note here that positive NMP represents CO₂ production (net heterotrophy). Black and red arrows represent statistically significant negative and positive effects, respectively ($p < 0.05$). Numbers adjacent to arrows are the standardized effect sizes of each relationship. *** $p < 0.001$; ** $p < 0.01$; * $p < 0.05$. Arrow width is proportional to the effect strength. SEM goodness of fit was evaluated based on variance explained by each of the two models (R^2). A summary of statistical outputs from the SEM model is provided in Supplementary Information **Table S2**.

Discussion

In this study, we observed that CO₂ emissions did not differ significantly between reservoirs with divergent trophic states. Instead, our results show that the carbonate system has a great potential to regulate the effect of net metabolic production on CO₂ emissions. This result extends observations by Marcé et al. (2015) that DIC inputs from catchments and the subsequent carbonate buffering need to be considered as an important contributor to CO₂ supersaturation to low alkalinity systems which can switch from CO₂ sources to weak CO₂ sinks. Primary productivity by itself is insufficient for estimations of CO₂ emissions from reservoirs. Ignoring the carbonate buffering capacity, therefore, is likely to result in a systematic bias towards underestimating CO₂ emissions. Our model approach offers quantitative insight on the potential contribution of carbonate buffering to the magnitude and seasonal variability of CO₂ in lakes and reservoirs under climate change.

Estimates of CO₂ emissions

The two reservoirs were CO₂ sources to the atmosphere over the annual cycle. The estimated CO₂ emissions from both reservoirs (62.3 and 67.9 g C m⁻² yr⁻¹) confirm estimates from Halbedel and Koschorreck (2013) (74.9 and 87.6 g C m⁻² yr⁻¹), which are based on much less data. Our estimates are similar to those of lake CO₂ emissions in temperate latitudes (Marcé et al., 2015), and to the overall magnitude and seasonal fluctuation of CO₂ emission from 39 German drinking reservoirs (Saidi and Koschorreck, 2017).

To assess the influence of sampling interval during estimates of annual CO₂ budgets in reservoirs, we randomly subsampled our biweekly dataset to simulate monthly, bimonthly, and seasonally sampling and compared the reconstructed total annual CO₂ fluxes to the original ones (**Fig. S7**). Artificially reducing our data density to monthly samplings resulted in virtually similar annual CO₂ emissions with the smallest variations (26% in Hassel and 20% in Rappbode).

Bimonthly sampling increased the error to 68% and 47% in both reservoirs, while seasonal-interval sampling will double the uncertainty. We therefore conclude that a bimonthly sampling frequency is probably an acceptable compromise between sampling effort and accuracy of the annual CO₂ budget.

We believe that omitting flood events would have little effect on our estimates of annual CO₂ emissions. We did not find substantial variations in CO₂ concentrations and emissions before, during, and after floods (**Fig. S8**), implying CO₂ chemostasis behavior (Aho et al., 2021). However, diffusive flux measurements should ideally be made frequently following ice out and then on a regular basis throughout the season until after mixing occurs in the fall (Wik et al., 2016).

We consider the uncertainty of k values has limited impact on the estimates of CO₂ emissions. CO₂ emissions were more regulated by CO₂ concentrations, rather than k (**Fig. S4**). Moreover, the range of wind speed at 1.5 m height typically varied within 2.0-3.3 m s⁻¹ (1st-3rd quartile), resulting in a relatively narrow range of k variations (0.49-1.39 m d⁻¹).

Variations in surface area had little effect on CO₂ emission estimates or subsequent model results. Flood events rarely fluctuate the water levels due to surface-release dams (Wendt-Potthoff et al., 2014), and high banks prevent the occurrence of bank overflow at the conjunction parts between inflows and reservoirs (Leng et al., 2021). 90% of the water level data varied within 0.1m (**Fig. S9**), resulting in expansion or shrinkage in the surface area by 2.0% and 2.4% of the total surface area in Hassel and Rappbode, respectively.

Regarding spatial heterogeneity, both monitoring sites are representative of entire reservoirs as shown by direct CO₂ emission measurements using floating chambers along the reservoirs in 2015 (**Table S3**) (coefficient of variance among different sampling sites less than |0.35|), except the conjunction parts to the inflows which represent 17% and 12% of the total area of Hassel and Rappbode Reservoirs, respectively.

Contributions of processes to CO₂ emissions

CO₂ in reservoirs originates either from direct CO₂ inputs (inflows and groundwater), in-reservoir metabolism, or carbonate buffering (from ionic forms of carbonate to CO₂). The low contribution of inflow CO₂ in our study can be explained by the pH levels in both reservoirs, since DIC contributes primarily in the form of bicarbonate, instead of CO₂ as other studies (Wilkinson et al., 2016). Influence of CO₂ derived from the catchments was highest in spring probably due to high hydraulic loading and shorter residence time.

NMP explained most of the seasonal variations in CO₂ dynamics in both reservoirs (**Fig. 4** and **Fig. S5e, f**). The high-nutrient reservoir showed higher productivity during summer (more negative NMP), leading to an autotrophic system (Bogard and Del Giorgio, 2016; Halbedel and Koschorreck, 2013). Besides that, thermal stratification existed during this period (**Fig. S10**), which results in CO₂ depletion at the surface and CO₂ accumulation in the hypolimnion (Wendt-Potthoff et al., 2014). The accumulation of CO₂ in the hypolimnion is not important to NMP in the epilimnion, especially in eutrophic and deeper lakes (Brothers et al., 2012; Vachon et al., 2017), as exchange across the thermocline is limited (Kreling et al., 2014). When the water column becomes colder in autumn, however, NMP increased drastically in both reservoirs, coinciding with increased water column turnover and the associated mobilization of hypolimnetic CO₂ (Vachon and Del Giorgio, 2014). Increased external DOC loading during autumn could be another potential source supporting pelagic respiration during autumn (**Fig. 2e**), dominating the net metabolic production (Morling et al., 2017; Vachon et al., 2017). This is nicely confirmed by the positive connection between DOC and NMP from the SEM results (**Fig. 6**) and the simulated ecosystem metabolism results.

The comparison of NMP from the mass balance model with the estimated CO₂ production (-NEP) from LakeMetabolizer allows us to validate the reliability of our model. The coupling between the two approaches showed that the NMP mostly resulted from -NEP while other biotic processes had limited impact on NMP. The metabolism model yields slightly lower estimate than the mass balance model (**Fig. 5c**, slope = -0.77, R² = 0.43, *p* = 0.006). This can theoretically be attributed to additional biochemical and photochemical reactions included in NMP. However, photochemical CO₂-production was probably of minor importance since it only contributed 3% to total CO₂ production in boreal lakes (Alleson et al., 2021). Anaerobic processes in the epilimnion can be ruled out in our case given the fact that no changes of solutes were observed in water samples from both reservoirs incubated under in situ temperature and O₂ conditions (Dadi et al., 2023).

The uncertainty resulting from the chosen respiratory quotient to convert the unit of -NMP from mmol O₂ m⁻² d⁻¹ to mmol CO₂ m⁻² d⁻¹, in our opinion, may be more essential in evaluating the comparability of NMP and -NEP (Berggren et al., 2012; Trentman et al., 2023). Compared to a reported respiratory quotient of 1.2 (Berggren et al., 2012), the quotient of 1:1 we used would underestimate the CO₂ production from metabolic processes. However, there was a large day-to-day variation in -NEP (**Fig. 5d**), reflecting the large variability in average daily CO₂ production in the epilimnion. Overall, the excellent agreement between mass balance and metabolism calculations underlines the reliability of our approach

under the studied time scale, whereas differences in chosen quotient and gas exchange between O₂ and CO₂ can generate a decoupling between both gases (Vachon et al., 2020).

Our mass balance model suggests a clear effect of the carbonate system on the size of the reservoir CO₂ pool. For both reservoirs, we found seasonal variations in CO₂ losses to and gains from carbonates. In addition, carbonate buffering can cause CO₂ production without O₂ consumption, uncoupling departure of O₂ and CO₂ (deviation from the 1:-1 line in **Fig. S3**). For both reservoirs, high positive carbonate buffering was observed in late spring and early summer along with the peak of external inorganic carbon contribution (**Fig. S5c-f**). This mechanism could be attributed to the ready replacement of evading and consumed CO₂ by external-derived carbonates. This leads to the coincidence of lowest net metabolic production and highest carbonate buffering (**Fig. 5e, f**), consequently attenuated seasonal variations in CO₂. Likewise, in autumn, when metabolic activity produces CO₂, the external carbonate source was able to shift carbonate equilibria from CO₂ towards carbonates, resulting in lower CO₂ emissions. During summer (~220-260 day of year), not only was stream discharge low, but also was the mixing layer depth (**Fig. S10**). This results in small absolute amounts of metabolic production and carbonate buffering but comparatively large importance of carbonate buffering (**Fig. S5**). Seasonal temperature increase can also lower the solubility of CO₂, eventually altering CO₂ gradient across the water-air interface. Apart from high alkalinity systems, which might cause seasonal variations in CO₂ emissions to be independent of NEP (Marcé et al., 2015), we emphasize that even in system with moderate alkalinity, the carbonate buffering capacity needs to be considered in seasonal CO₂ budgets, notably during summer, when it can contribute up to 50% of CO₂ emissions.

The estimated reservoir CO₂ budgets provide evidence how in-reservoir metabolic processes and carbonate buffering modulate the seasonal variations in CO₂ evasion in low-alkalinity systems. The mass balance model developed by Vachon et al. (2017) considered the shifting pH brought by CO₂ production and consumption, and hence DIC speciation equilibrium, to play a minimal impact. This is reasonable in lakes and reservoirs with low alkalinity/pH (Soued and Prairie, 2021; Vachon et al., 2017). In high pH reservoirs the mass-balance models took the extra carbonate precipitation into account (Finlay et al., 2010; López et al., 2011). Our study complements the findings from Marcé et al. (2015) that highlighted DIC generated from catchment weathering could potentially account for half of the emissions in temperate latitudes. McDonald et al. (2013) and Weyhenmeyer et al. (2015) documented the widespread DIC loading in determining lake CO₂, and pointed out the necessity to consider the temporal variability in hydrologic DIC inputs. Our findings reveal that the external DIC load

and carbonate buffering are not always coupled over time. Alternatively, once the reservoirs receive external inputs, the exchange between carbonates and CO₂ would only take place when there was a difference in alkalinity between the inflow and receiving water – that is especially in high pH situations.

Implications to current CO₂ budgets and future predictions

Reservoirs are currently confronted by a wide array of changes due to climate change, posing a threat to water resource security as well as changing C cycling. Altered rainfall pattern are affecting input of nutrients into reservoirs. Indirect effects of climate change, such as deforestation, as happening in the Rappbode catchment and resulting in 17% of forest loss in 2020 alone (Kong et al., 2022), lead to enhanced nutrient export from the catchment to reservoirs through surface runoff and leaching processes. Our SEM results indicate that from the seasonal perspective, drought not only raises in-reservoir DOC, leading to higher metabolic CO₂ production, but also increases external DIC import. Reservoir productivity is highest when light and drought coincide (summer) (Halbedel and Koschorreck, 2013). The general pattern of CO₂ regulation seems to be relatively independent of trophic state (**Fig. 3** and **Fig. S4**). Thus, classifying reservoirs according to catchment alkalinity might be more relevant for predicting CO₂ emissions than looking at trophic state.

We strongly recommend including carbonate buffering and its response to environmental change in future scenarios for predicting the effects of global change in lentic systems with low alkalinity. This is because the carbonate system is much more sensitive to the changes in in-reservoir processes than in high alkalinity systems, where carbonate buffering would behave more independent of NEP. Previous studies showed that alkalinity in more than 50% of global reservoirs and lakes is less than 1 meq L⁻¹ (Marcé et al., 2015). Consideration of the offsetting effect of carbonate buffering will largely reduce the uncertainty of global inland water CO₂ emission estimations. In our reservoirs, we show that ruling out the link between CO₂ emissions and external DIC input would result in up to 72% of unexplainable CO₂ emissions. Concentrations of nutrients and inorganic carbon in inflows reach the peak for the year simultaneously (**Fig. S2**), thus the eutrophication (consumes CO₂ due to larger algal production) and allochthonous inorganic carbon input (shifts in carbonate equilibrium) impact CO₂ emissions concurrently in opposite directions (**Fig. 1a**).

Conclusions

Our process-based mass balance model approach based on the large data set collected over multiple years in two reservoirs suggests that not only the effects of elevated nutrient loads and eutrophication on CO₂ need to be considered.

690 Changed loads of DIC may exhibit an equally important effect on reservoir CO₂
691 emissions. Consistent with our hypothetical responses of reservoir CO₂ emissions
692 (**Fig. 1b**), our study indicates that (1) each process influencing CO₂ budgets
693 varies seasonally in an opposite direction, explaining a more conservative
694 seasonal CO₂ dynamics, and (2) current models of reservoir CO₂ dynamics that
695 are based solely on internal metabolism miss a substantial portion of CO₂
696 produced from external DIC inputs.

697 Our model approach should also be applicable in determining the carbon
698 budgets in natural lakes beyond our study sites, because the principles regulating
699 carbon cycling in different aquatic systems would be highly similar. Even though
700 further research is still needed to incorporate the fate of terrestrial carbon and in-
701 reservoir metabolic processing, we recommend particularly the inclusion of
702 carbonate buffering into carbon budgets, to reduce uncertainty in predicting lake
703 and reservoir CO₂ emissions under global change.

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